SAFETY DATA SHEETS

According to Globally Harmonized System of Classification and Labelling of Chemicals (GHS) - Sixth revised edition

Version: 1.0 Creation Date: Aug 12, 2017 Revision Date: Aug 12, 2017

1.Identification

1.1 GHS Product identifier Product name cetyltrimethylammonium chloride 1.2 Other means of identification Product number Hexadecyltrimethylammonium chloride Other names 1.3 Recommended use of the chemical and restrictions on use Identified uses Uses advised against For industry use only. Adhesives and sealant chemicals, Intermediates, Surface active agents no data available 1.4 Supplier's details Company Address Echemi.com Echemi.com Echemi.com Echemi.com Telephone Fax 1.5 Emergency phone number Emergency phone number Echemi.com Service hours Monday to Friday, 9am-5pm (Standard time zone: UTC/GMT +8 hours). 2.Hazard identification

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2.1 Classification of the substance or mixture

Acute toxicity - Oral, Category 4 Acute toxicity - Dermal, Category 3 Skin corrosion, Category 1C Serious eye damage, Category 1 Hazardous to the aquatic environment, short-term (Acute) - Category Acute 1 Hazardous to the aquatic environment, long-term (Chronic) - Category Chronic 1

2.2 GHS label elements, including precautionary statements

Pictogram(s)

Signal word	Danger
Hazard statement(s)	H302 Harmful if swallowed
	H311 Toxic in contact with skin
	H314 Causes severe skin burns and eye damage
	H410 Very toxic to aquatic life with long lasting effects
Precautionary statement(s)	
Prevention	P264 Wash thoroughly after handling.
	P270 Do not eat, drink or smoke when using this product.
	P280 Wear protective gloves/protective clothing/eye protection/face protection.
	P260 Do not breathe dust/fume/gas/mist/vapours/spray.
	P273 Avoid release to the environment.
Response	P301+P312 IF SWALLOWED: Call a POISON CENTER/doctor/if you feel unwell.
	P330 Rinse mouth.
	P302+P352 IF ON SKIN: Wash with plenty of water/
	P312 Call a POISON CENTER/doctor/if you feel unwell.
	P321 Specific treatment (see on this label).
	P361+P364 Take off immediately all contaminated clothing and wash it before reuse.
	P301+P330+P331 IF SWALLOWED: Rinse mouth. Do NOT induce vomiting.
	P303+P361+P353 IF ON SKIN (or hair): Take off immediately all contaminated clothing. Rinse skin with water [or shower].
	P363 Wash contaminated clothing before reuse.
	P304+P340 IF INHALED: Remove person to fresh air and keep comfortable for breathing.
	P310 Immediately call a POISON CENTER/doctor/
	P305+P351+P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.
	P391 Collect spillage.
Storage	P405 Store locked up.
Disposal	P501 Dispose of contents/container to

2.3 Other hazards which do not result in classification

none

3.Composition/information on ingredients

3.1 Substances

	Chemical name	Common names and synonyms	CAS number	EC number	Concentration
cetyltr	imethylammonium chloride	cetyltrimethylammonium chloride	112-02-7	none	100%

4.First-aid measures

4.1 Description of necessary first-aid measures

General advice

Consult a physician. Show this safety data sheet to the doctor in attendance.

If inhaled

If breathed in, move person into fresh air. If not breathing, give artificial respiration. Consult a physician.

In case of skin contact

Wash off with soap and plenty of water. Consult a physician.

In case of eye contact

Rinse thoroughly with plenty of water for at least 15 minutes and consult a physician.

If swallowed

Never give anything by mouth to an unconscious person. Rinse mouth with water. Consult a physician.

4.2 Most important symptoms/effects, acute and delayed

Ingestion may produce toxic effects. Contact with eyes or skin may cause severe damage. (USCG, 1999)

4.3 Indication of immediate medical attention and special treatment needed, if necessary

no data available

5.Fire-fighting measures

5.1 Extinguishing media

Suitable extinguishing media

...DRY CHEM, /ALCOHOL/ FOAM, OR CARBON DIOXIDE. WATER MAY BE INEFFECTIVE ON FIRE. COOL EXPOSED CONTAINERS WITH WATER. /CETYLTRIMETHYLAMMONIUM CHLORIDE SOLN/

5.2 Specific hazards arising from the chemical

Special Hazards of Combustion Products: Irritating fumes of hydrogen chloride may form in fires. Behavior in Fire: Solvent vapors are heavier than air and may travel to a source of ignition and flash back. (USCG, 1999)

5.3 Special protective actions for fire-fighters

Wear self-contained breathing apparatus for firefighting if necessary.

6.Accidental release measures

6.1 Personal precautions, protective equipment and emergency procedures

Use personal protective equipment. Avoid dust formation. Avoid breathing vapours, mist or gas. Ensure adequate ventilation. Evacuate personnel to safe areas. Avoid breathing dust. For personal protection see section 8.

6.2 Environmental precautions

Prevent further leakage or spillage if safe to do so. Do not let product enter drains. Discharge into the environment must be avoided.

6.3 Methods and materials for containment and cleaning up

Pick up and arrange disposal. Sweep up and shovel. Keep in suitable, closed containers for disposal.

7.Handling and storage

7.1 Precautions for safe handling

Avoid contact with skin and eyes. Avoid formation of dust and aerosols. Avoid exposure - obtain special instructions before use. Provide appropriate exhaust ventilation at places where dust is formed. For precautions see section 2.2.

7.2 Conditions for safe storage, including any incompatibilities

Store in cool place. Keep container tightly closed in a dry and well-ventilated place.

8.Exposure controls/personal protection

8.1 Control parameters

Occupational Exposure limit values

no data available

Biological limit values

no data available

8.2 Appropriate engineering controls

Handle in accordance with good industrial hygiene and safety practice. Wash hands before breaks and at the end of workday.

8.3 Individual protection measures, such as personal protective equipment (PPE)

Eye/face protection

Safety glasses with side-shields conforming to EN166. Use equipment for eye protection tested and approved under appropriate government standards such as NIOSH (US) or EN 166(EU).

Skin protection

Wear impervious clothing. The type of protective equipment must be selected according to the concentration and amount of the dangerous substance at the specific workplace. Handle with gloves. Gloves must be inspected prior to use. Use proper glove removal technique(without touching glove's outer surface) to avoid skin contact with this product. Dispose of contaminated gloves after use in accordance with applicable laws and good laboratory practices. Wash and dry hands. The selected protective gloves have to satisfy the specifications of EU Directive 89/686/EEC and the standard EN 374 derived from it.

Respiratory protection

Wear dust mask when handling large quantities

Thermal hazards

no data available.

9. Physical and chemical properties

Physical state Colour Odour Melting point/ freezing point Boiling point or initial boiling point and boiling range Flammability white crystalline powder no data available no data available 232-234°C 100°C

 Lower and upper explosion limit / flammability limitIN AIR: 2%-12% /ISOPROPYL ALCOHOL SOLN OF CETYLTRIMETHYLAMMONIUM CHLORIDE/

 Flash point
 20.56°C (USCG, 1999)

 Auto-ignition temperature
 398.89°C (USCG, 1999)

 Decomposition temperature
 no data available

 pH
 no data available

 Solubility
 no data available

 Partition coefficient n-octanol/water (log value)
 log Kow= 3.23

 Vapour pressure
 no data available

 Density and/or relative density
 0.968g/mLat 25°C

 Relative vapour density
 no data available

10.Stability and reactivity

10.1 Reactivity

no data available

10.2 Chemical stability

Stable under recommended storage conditions.

10.3 Possibility of hazardous reactions

HEXADECYLTRIMETHYLAMMONIUM CHLORIDE is a quaternary ammonium compound. Therefore it has a permanently charged nitrogen atom with four alkyl or aryl groups on it. Compounds in this group are generally soluble in water and are near pH 7 (neutral). They do not undergo the acid-base reactions typical of primary, secondary, or tertiary ammonium ions. Quaternary ammonium salts often serve as catalysts in reactions. They are incompatible with many strong oxidizers and reducing agents, such as metal hydrides, alkali/active metals, and organometallics.

10.4 Conditions to avoid

no data available

10.5 Incompatible materials

no data available

10.6 Hazardous decomposition products

When heated to decomposition it emits very toxic fumes of /hydrogen chloride and nitrogen oxides/.

11.Toxicological information

Acute toxicity

- · Oral: no data available
- Inhalation: no data available
- Dermal: no data available

Skin corrosion/irritation

no data available

Serious eye damage/irritation

no data available

Respiratory or skin sensitization

no data available

Germ cell mutagenicity

no data available

Carcinogenicity

no data available

Reproductive toxicity

no data available

STOT-single exposure

no data available

STOT-repeated exposure

no data available

Aspiration hazard

no data available

12. Ecological information

12.1 Toxicity

- · Toxicity to fish: no data available
- · Toxicity to daphnia and other aquatic invertebrates: no data available
- Toxicity to algae: no data available
- · Toxicity to microorganisms: no data available

12.2 Persistence and degradability

THE CATIONIC SURFACTANTS BIODEGRADATION IN SEWAGE AND RIVER WATER WAS STUDIED, MEASURING THE REDUCTION OF THE ACTIVE SUBSTANCE BY A COLORIMETRIC METHOD USING SODIUM ALIZARINE SULFONATE.

12.3 Bioaccumulative potential

An estimated BCF of 71 was calculated for trimethylhexadecylammonium chloride(SRC), using an estimated log Kow of 3.23(1) and a regression-derived equation(2). According to a classification scheme(3), this BCF suggests the potential for bioconcentration in aquatic organisms is moderate.

12.4 Mobility in soil

Using a structure estimation method based on molecular connectivity indices(1), the Koc for trimethylhexadecylammonium chloride can be estimated to be 2.2X10+5(SRC). According to a classification scheme(2), this estimated Koc value suggests that trimethylhexadecylammonium chloride is expected to immobile in soil.

12.5 Other adverse effects

no data available

13.1 Disposal methods

Product

The material can be disposed of by removal to a licensed chemical destruction plant or by controlled incineration with flue gas scrubbing. Do not contaminate water, foodstuffs, feed or seed by storage or disposal. Do not discharge to sewer systems.

Contaminated packaging

Containers can be triply rinsed (or equivalent) and offered for recycling or reconditioning. Alternatively, the packaging can be punctured to make it unusable for other purposes and then be disposed of in a sanitary landfill. Controlled incineration with flue gas scrubbing is possible for combustible packaging materials.

14.Transport information

14.1 UN Number		
ADR/RID: UN3082	IMDG: UN3082	IATA: UN3082
14.2 UN Proper Shipping Name		
ADR/RID: ENVIRONMENTALLY HAZARDOU: IMDG: ENVIRONMENTALLY HAZARDOUS SI IATA: ENVIRONMENTALLY HAZARDOUS SU	UBSTANCE, LIQUID, N.O.S.	
14.3 Transport hazard class(es)		
ADR/RID: 9	IMDG: 9	IATA: 9
14.4 Packing group, if applicable		
ADR/RID: III	IMDG: III	IATA: III
14.5 Environmental hazards		
ADR/RID: yes	IMDG: yes	IATA: yes
14.6 Special precautions for user		

no data available

14.7 Transport in bulk according to Annex II of MARPOL 73/78 and the IBC Code

no data available

15.Regulatory information

15.1 Safety, health and environmental regulations specific for the product in question

Chemical name	Common names and synonyms	CAS number	EC number
cetyltrimethylammonium chloride	cetyltrimethylammonium chloride	112-02-7	none
European Inventory of Existing Commercial Chemical Substances (EINECS)			Listed.
EC Inventory			Listed.
United States Toxic Substances Control Act (TSCA) Inventory			Listed.
China Catalog of Hazardous chemicals 2015			Not Listed.
New Zealand Inventory of Chemicals (NZIoC)			Listed.
Philippines Inventory of Chemicals and Chemical Substances (PICCS)			Listed.
Vietnam National Chemical Inventory			Listed.
Chinese Chemical Inventory of Existing Chemical Substances (China IECSC)			Listed.

16.Other information

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Information on revision	
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- Abbreviations and acronyms
 - CAS: Chemical Abstracts Service
 - ADR: European Agreement concerning the International Carriage of Dangerous Goods by Road RDD: Regulation concerning the International Carriage of Dangerous Goods by Rail

 - MDG: International Maritime Dangerous Goods
 IATA: International Air Transportation Association
 TWA: Time Weighted Average
 STEL: Short term exposure limit
 LCS0: Lethal Concentration 50%

 - LD50: Lethal Dose 50% EC50: Effective Concentration 50%

References

- IPCS The International Chemical Safety Cards (ICSC), website: http://www.ilo.org/dyn/icsc/showcard.home
- · HSDB Hazardous Substances Data Bank, website: https://toxnet.nlm.nih.gov/newtoxnet/hsdb.htm
- · IARC International Agency for Research on Cancer, website: http://www.iarc.fr/
- eChemPortal The Global Portal to Information on Chemical Substances by OECD, website: http://www.echemportal.org/echemportal/index?pageID=0&request_locale=en
- CAMEO Chemicals, website: http://cameochemicals.noaa.gov/search/simple
- ChemIDplus, website: http://chem.sis.nlm.nih.gov/chemidplus/chemidlite.jsp
- ERG Emergency Response Guidebook by U.S. Department of Transportation, website: http://www.phmsa.dot.gov/hazmat/library/erg
- Germany GESTIS-database on hazard substance, website: http://www.dguv.de/ifa/gestis/gestis-stoffdatenbank/index-2.jsp
- ECHA European Chemicals Agency, website: https://echa.europa.eu/

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